INTRODUCTION

Chitosan is a deacetylated form of chitin, a polymer of heteropolysaccharide and composed of N-acetyl glucosamine and 70 % or more N-glucosamine. Chitooligosaccharides, obtained from chitosan by hydrolysis with chitosanase, have various potential applications in biomedicines, pharmaceuticals, agriculture and food. Chitosanase (EC 3.2.1.132) is a glycoside hydrolase which catalyzes the hydrolysis of β-1,4-linkages between N-acetyl-D-glucosamine and D-glucosamine residues in a partly acetylated chitosan. This enzyme has specific hydrolysis pattern which depends on the source of microorganisms, including bacteria and fungi.

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In an effort of obtaining indigenous chitosanase producing bacteria, bacterial isolates were collected from Indralaya swamp, South Sumatera. The extracellular chitosanase-production capacity of 14 bacterial isolates taken from Indralaya Swamp South Sumatera, was evaluated in liquid cultures. The GB2 isolate was selected due to its high chitosanase activity. Therefore, the objective of this study was to characterize chitosanase from bacteria Indralaya swamp, South Sumatera.

EXPERIMENTAL

GB2 isolate used in this experiment was isolated from swamp in Indralaya South Sumatera. Media production : 0.5 % colloidal citin; 0.1 % K$_2$HPO$_4$; 0.01 % MgSO$_4$.7H$_2$O; 0.05 % yeast extract; 0.1 % NaCl and 0.7 % (NH$_4$)$_2$SO$_4$.

Enzyme production: Enzyme production was determined based on daily analyses of cell free broth, for chitosanase activity. Incubation was conducted at 37 ºC with agitation 120 rpm.

Chitosanase activity: Chitosanase assay was conducted according to Yoon et al. with modification. The reaction mixture consisting of 100 µL of 1 % soluble colloidal chitosan, 100 µL phosphate buffer 0.05 M (pH 6) and 100 µL of the enzyme solution was incubated at 37 ºC or at optimum temperature for 30 min with shaking. The reaction was stopped by incubating the mixture at -10 ºC for 15 min. The amount of reducing sugar in the mixture was determined by a modified method of Schales with glucosamine as a reference product.

An amount of 200 µL of the mixture solution was further mixed with 1 mL Schales reagent and 800 µL aquadest. After covered with aluminum foil, the tubes were heated in boiling water for 15 min, centrifuged for 10 min at 3000 g and the absorbance was read at $\lambda = 420$ nm. A blank was prepared using aquadest instead of sample solution. One (1) unit of chitosanase activity was defined as the amount of the enzyme which produces 1 µmol of reducing sugar (glucosamine) per minute.

Protein determination: Protein content was determined based on Bradford method using bovine serum albumin as the standard at 0.2-1.2 mg protein/mL stock bovine serum albumin. Following Guttenberger procedure, the reaction mixture containing 100 µL of sample, 1 mL of aquadest and 1 mL Bradford reagent. After vortexing the mixture, the absorbance was read at 595 nm. A blank was prepared by substituting sample solution with 100 µL of aquadest.
Effect of pH and temperature on enzyme activity:

Optimum pH was determined by assaying in buffer with pH values of 6-9 using 0.05 M universal buffer composed of citric acid, KH$_2$PO$_4$, diethylbarbituric acid and 0.2 N NaOH, in the presence of soluble chitosan substrate.

Effect of metal Ions on enzyme activity: To determine the effect of metal ions, various cation monovalent such as Na$^+$, K$^+$ and cation divalent such as Ca$^{2+}$ and Fe$^{2+}$ at concentration of 1 mM were added and incubated for 10 min. At 37 °C before assay the enzyme activity. The residual activity was presented as % relative activity based on control (enzyme activity without addition of metal ions).

Molecular weight determination: Molecular weight was measured by electrophoresis under denaturing polyacrylamide-SDS. Standard protein of low molecular weight marker was used for molecular weight determination. Enzymes activity in situ was determined by zymogram following previous reports modifications. SDS-PAGE was performed in 10 % acrylamide. For zymogram analysis, soluble chitosan at 0.1 % was incorporated into the gel.

RESULTS AND DISCUSSION

Enzyme production: GB2 isolate grew well in the modified medium of Park et al.$^{10}$ containing 0.1 % K$_2$HPO$_4$; 0.01 % MgSO$_4$.7H$_2$O; 0.05 % yeast extract; 0.1 % NaCl and 0.7 % (NH$_4$)$_2$SO$_4$. Fig. 1 show the optimum fermentation time. The optimum production of GB2 chitosanases was 3 day incubation.

Effect of temperature on enzyme activity: The optimum temperature of GB2 chitosanase was 30 °C. Fig. 2 show the effect of temperature on chitosanase activity from swamp bacteria.

Effect of metal Ions on enzyme activity: Effect of metal ions, Ca$^{2+}$ and Fe$^{2+}$ were the activator whereas K$^+$ and Na$^+$ inhibited GB2 chitosanase. Fig. 4 showed effect of metal ions on chitosanase activity.
Molecular weight determination: Molecular weights were determined by using SDS-PAGE and zymogram technique. Molecular weight chitosanase isolate GB2 was 36-110 kD (Fig. 5).

GB2 isolate grew well in the modified medium of Park et al. containing 0.5 % colloidal citin; 0.1 % K$_2$HPO$_4$; 0.01 % MgSO$_4$·7H$_2$O; 0.05 % yeast extract; 0.1 % NaCl and 0.7 % (NH$_4$)$_2$SO$_4$. The optimum enzyme production was 3 day of incubation (Fig. 3). This findings was rather different with previous study on production of chitosanase from isolate 96, 97, 99, 100 and Matsuebacter chitosanotabidus 3001. The optimum production of chitosanase from isolate 97, 99, 100 and Matsuebacter chitosanotabidus 3001 were 5 day incubation and isolate 96 was 4 day incubation.

The optimum pH of chitosanase under assay condition was 8 (Fig. 2). Optimum pH of chitosanase from bacteria and fungi were in range 4-8 and that optimum pH from isolate GB2 was within the range. Chitosanase molecule is cationic, therefore electrostatic interactions is believed to contribute in the substrate binding significantly. Study on many chitosanase reported that aspartate was important in substrate binding, while glutamate, in addition to aspartate, were involved in catalysis reaction. Optimum temperature for GB2 chitosanase was 30 ºC. A chitosanase which is active in 30 ºC has also catalysis reaction as indicated by the clear bands (Fig. 5).

Fig. 4 showed Ca$^{2+}$ and Fe$^{3+}$ as activator whereas Na$^+$ and K$^+$ inhibited chitosanase GB2. Positively charged ions such as metal ions generally bind to carboxylate, imidazole and sulfhydryl groups of the side chains of the amino acids. This might imply the important of negatively charged amio acid such as aspartat and glutamate for catalytic activity. The presence of metal cations is needed to obtain local charge balance of the catalytic site. In general, metal ions can function as part of the enzyme catalytic mechanism and or stabilizer of the active site of the protein enzyme. When participating in the catalytic mechanism, metal cation can bring specific functional group together in the appropriate orientation for reaction. The different effect of cation upon catalytic action of the enzyme may be due to their ability to adopt different geometrics in the same site in the absence of substrate.

SDS-PAGE and zymogram analysis was performed to estimate the molecular weight. SDS-PAGE analysis of chitosanase GB2 was in the range of 14.1-110 kD. The molecular weight of chitosanase GB2 was estimated 36-110 kD as indicated by the clear bands (Fig. 5).

Conclusion

The GB2 isolate was selected due to its high chitosanase activity. The optimum pH and temperature of chitosanase from GB2 were 8 and 30 ºC, respectively. Effect of metal ions study, showed that Ca$^{2+}$ and Fe$^{3+}$ were activator whereas Na$^+$ and K$^+$ were acted as inhibitor to chitosanase GB2. Study on the effect of metals ion indicated that chitosanase GB2 was metalloenzyme. Molecular weights were determined by using SDS-PAGE and zymogram technique. Molecular weight chitosanase isolate GB2 was estimated around 36-110 kD.

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